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Schinalactone A, a New Cytotoxic Triterpenoid from *Schisandra sphenanthera*

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ABSTRACT

A new cytotoxic triterpenoid, schinalactone A (1), together with two new biogenetically related compounds, schinalactones B (2) and C (3),



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During the past 10 years, our group has made great efforts in the study of the chemical constituents and biological activities of the genus *Schisandra*, resulting in the discovery of more than 100 highly oxygenated nortriterpenoids with nine unprecedented carbon skeletons.¹

Those unusual ring assemblies highly oxygenated structures have brought great interest and challenges to chemists for total synthesis and biogenetic studies.² The previous work carried out by our group on nortriterpenoids of *Schisandra sphenan*-*thera* Rehd et Wils led to finding a series of novel nortriterpenoids with complicated polycyclic systems.³ In a continuing search for structurally unique and biogenetically interesting triterpenoids, the chemical constituents of *S. sphenanthera*

has been isolated from the roots and stems of *Schisandra sphenanthera*. Their structures were elucidated on the basis of extensive spectroscopic analysis. Compounds 1 and 2 showed significant cytotoxicity against PANC-1 cell lines with IC_{50} values of 5.9 and 4.1 μ M, respectively. A plausible biosynthetic pathway of 1 was also postulated.

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triterpenoids, schinalactones B (2) and C (3), and one known compound, lancifoic acid A (4), had been isolated from the roots and stems of this plant. Among them, 1 possessed a rearranged cycloartane type triterpenoid skeleton with a novel five-membered carbon ring featuring C-30 connected to C-1 (ring B) and showed significant cytotoxicity against SK-BR-3 and PANC-1 cell lines with IC₅₀ values of 5.2 and 5.9 μ M, respectively, which was one of the most significant cytotoxic triterpenoids isolated from the *Schisandra* species. In this paper, the isolation, structure elucidation, and biological activities of these new compounds are described.



Schinalactone A (1) was obtained as a white, amorphous powder, and the molecular formula $C_{30}H_{42}O_7$ was assigned through its HRESIMS (*m*/*z* 537.2839, [M + Na]⁺), requiring 10 degrees of unsaturation. The ¹H and ¹³C NMR spectra of 1 revealed the presence of two lactone carbonyl carbons and a pair of olefinic carbons. Apart from three degrees of unsaturation occupied by one double bond and two carbonyls, the remaining seven degrees of unsaturation indicated that 1 should possess a heptacyclic system. Comparison of the ¹H and ¹³C NMR data of 1 with those of kadcoccilactone A⁴ suggested that they were similar and might both be derived from cycloartane type triterpenoids and differed mainly in ring A and ring B. In 1, a six-membered lactone



Figure 1. Selected HMBC (H \rightarrow C) and ¹H $^{-1}$ H COSY (-) correlations of 1.

ring (ring A) was assigned on the basis of observations from HMBC (Figure 1) correlations of H-2 with C-1, C-3 and C-10, and of H-1 with C-2 and C-3. Furthermore, weak HMBC correlation from H₃-29 ($\delta_{\rm H}$ 1.30, 3H, s) to C-3 ($\delta_{\rm C}$ 172.4) also suggested that the C-3 connected to C-4 ($\delta_{\rm C}$ 90.6) through an oxygen bridge. In addition, a five-membered carbon ring fused system (ring B) was deduced from HMBC correlations from H-5 to C-1, C-10, and C-30, from H-1 to C-4 and C-30, and from H₂-30 to C-4 and C-5, coupled with ¹H⁻¹H COSY (Figure 1) correlations of H-1 with H-2 and H₂-30. Further evidence supporting fusion of rings A and B was provided by the presence of a spin system (H_2 -30/H-1/ H-2) as deduced from the COSY spectrum and HMBC correlation from H-2 to C-30. From the above deduction, coupled with the molecular formula and degrees of unsaturation, there must be one more ring. From the HMBC spectrum, two oxygenated quaternary carbons C-9 and C-10 were assigned on the basis of correlations from H-7, H-8, H-11, and H-19 to C-9 and from H-2, H-6, and H-19 to C-10, respectively. Meanwhile, the positive FAB mass spectrum of 1 revealed a moderately strong fragment peak at m/z 465 $[M + H - H_2O - O_2]^+$, indicating the cleavage of a peroxyl group. Moreover, an AB doublet for the C-19 methylene group resonances occurred at relatively low field region ($\delta_{\rm H}$ 2.88, 1H, d, J = 12.2 Hz; $\delta_{\rm H}$ 2.26, 1H, d, J = 12.2 Hz) due to the effect of the peroxyl bridge.⁵ These data indicated that the peroxyl bridge was located between C-9 and C-10. Consequently, the planar structure of 1 was assigned by comprehensive analysis of 2D NMR data, including HSQC, HSOC-TOCSY, ¹H-¹H COSY, and HMBC experiments. The planar structure of 1 was further confirmed by comparing the ¹³C NMR chemical shift differences between 1 and its acetate derives (1a) (Scheme 1) as well. Chemical shift differences ($\Delta \delta_{\rm C} = \delta_{\rm C} \mathbf{1} - \mathbf{1} \mathbf{a}$) between 1 and 1a of C-1 ($\Delta \delta_{\rm C}$ +2.0 ppm), C-3 ($\Delta \delta_{\rm C}$ +5.3 ppm), C-4 ($\Delta \delta_{\rm C}$ -0.9 ppm), and C-30 ($\Delta \delta_{\rm C}$ –0.4 ppm) indicated that these carbons were proximal to C-2.

With the gross structure of **1** in hand, the relative stereochemistry of **1** was readily assigned by ROESY correlations and modeling in Chem3D 10.0 (Cambridge Soft, Inc.) as shown in Figure 2. Correlations between H₂-19 β ($\delta_{\rm H}$

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2.88) with H-6 β and H-8 β clearly showed that these protons were on the same face. Thus, the peroxyl bridge in 1 was deduced to be in the α -orientation. The lactone ring between C-3 and C-4 adopted an α -orientation, which was deduced from strong correlations of H₂-30/H-1 and H-30 β /H-6 β . On the contrary, correlation of H-30 β /H-6 β in the ROESY spectrum might have disappeared when the lactone ring (ring A) adopted the β -orientation on C-4. In addition, a broad singlet at $\delta_{\rm H}$ 4.85 ppm (H-2) in the ¹H NMR spectrum indicated that the dihedral angle between H-1 and H-2 should be 90°, so that the hydroxyl at C-2 should have adopted a β -orientation. H-22 has strong correlation with H-16 β in ROESY spectrum, indicating that H-22 was in the α -orientation. The relative configuration of other stereocenters in 1 was the same as kadcoccilactone A. On the basis of the above deduction, the carbon backbone of 1 was unambiguously established.

Schinalactone B (2) was deduced to have the molecular formula $C_{30}H_{46}O_5$ from its HRESIMS at m/z 509.3251 ([M + Na]⁺, calcd 509.3243), requiring 8 degrees of unsaturation. On analysis of its ¹H and ¹³C NMR spectra, features similar to those of schisanterpene A⁶ were evident, but a pair of double bonds between C-4 and C-30 in schisanterpene A was absent in **2**, which instead exhibited a tertiary alcohol (δ_C 72.7). As a result, **2** was pentacyclic and belonged to 3,4;9,10-*seco*-cycloartane-type triterpenoids. The oxygen atom of the spiro-ring on C-10 was in the α -orientation, which was determined by ROESY correlations (between H-2 β and H₂-19 and H-1 β and H₃-29). H-22 has strong correlation with H-16 β in the ROESY spectrum, indicating that H-22 was in the α -orientation.

Schinalactone C (3) was isolated as colorless needles and exhibited a $[M + Na]^+$ ion peak at m/z 509.3245 in the

Table	1.	^{13}C	NMR	Data	of	Com	pounds	1-2	3 in	Pvridi	$1e-d_5^a$
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Table 1.	C NMR Data of Compounds 1 5 in Fyndinc-45						
	1	1a	2	3			
1	48.5 (d)	46.5 (d)	29.1 (t)	31.1 (t)			
2	67.4 (d)	67.8 (d)	30.3 (t)	32.0 (t)			
3	172.4~(s)	167.1 (s)	177.5(s)	174.8(s)			
4	90.6 (s)	91.5 (s)	72.7~(s)	75.1(s)			
5	59.7 (d)	59.2 (d)	55.7 (d)	45.9 (d)			
6	27.1(t)	27.0 (t)	25.6 (t)	25.5(t)			
7	26.7 (t)	26.6 (t)	33.5 (t)	26.2 (t)			
8	48.7 (d)	48.7 (d)	31.3 (d)	48.8 (d)			
9	86.9 (s)	87.2(s)	47.9 (d)	22.6(s)			
10	91.8 (s)	91.5 (s)	92.2 (s)	27.5~(s)			
11	35.5 (t)	35.2 (t)	26.7(t)	26.7(t)			
12	30.6 (t)	30.5 (t)	32.9 (t)	33.2 (t)			
13	45.7~(s)	45.6 (s)	46.1 (s)	45.5~(s)			
14	48.2(s)	48.2(s)	48.8 (s)	48.8(s)			
15	33.6 (t)	33.5 (t)	31.9 (t)	36.3 (t)			
16	26.6 (t)	26.4 (t)	33.4 (t)	27.1(t)			
17	46.6 (d)	46.1 (d)	47.1 (d)	48.4 (d)			
18	14.1 (q)	14.0 (q)	14.7 (q)	18.5 (q)			
19	57.2(t)	56.5 (t)	52.1 (t)	31.5 (t)			
20	39.5 (d)	39.5 (d)	39.5 (d)	39.4 (d)			
21	13.6 (q)	13.5 (q)	13.5 (q)	13.1(q)			
22	80.4 (d)	80.4 (d)	80.4 (d)	80.5 (d)			
23	23.8 (t)	23.6 (t)	23.6 (t)	23.6 (t)			
24	140.2 (d)	140.1 (d)	140.1 (d)	140.2 (d)			
25	128.0(s)	128.0(s)	128.0(s)	128.0(s)			
26	166.3 (s)	166.3 (s)	166.3(s)	166.3 (s)			
27	17.2~(q)	17.1 (q)	17.2~(q)	17.3 (q)			
28	18.0(q)	17.9 (q)	16.7 (q)	19.8 (q)			
29	21.6~(q)	21.4(q)	31.4 (q)	32.0 (q)			
30	35.6(t)	36.0 (t)	29.7 (q)	26.7 (q)			

^{*a*} Data for compounds **1**, **1a**, and **3** were recorded at 125 MHz, data for compound **2** were recorded at 100 MHz, and assignments were based on HSQC, COSY, HMBC, and ROESY experiments.

HRESIMS, associated with a molecular formula of $C_{30}H_{46}O_5$. The presence of a three-membered ring as suggested by characteristic signals in the ¹H and ¹³C NMR spectra ($\delta_{\rm C}$ 22.6, s, C-9; $\delta_{\rm C}$ 27.5, s, C-10; $\delta_{\rm C}$ 31.5, t, C-19, and $\delta_{\rm H}$ 0.58, ABd, J = 5.5 Hz, and 0.79, ABd, J = 5.5 Hz, H₂-19) indicated that 3 belonged to the cycloartane skeleton triterpenoid family (Table 1). The NMR data of 3 were very similar to those of lancifoic acid A (4),⁷ which belongs to 3,4-seco-cycloartane-type triterpenoids, with the main difference being that 3 is pentacyclic whereas 4 is tetracyclic. There is a six-membered α,β -unsaturated lactone ring in 3, which was supported by ¹H-¹H COSY and HMBC correlations. H-22 has strong correlation with H-16 β in ROESY spectrum, indicating that H-22 was on α -orientation. The other relative stereochemistry centers were identical to those of 4.

The co-occurrence of 1 with various 3,4-seco- or 3,4;9,10seco-cycloartane triterpenoids (2-4) within the same plant raises interesting questions about the biogenesis of 1. Biogenetically, 1 may be produced from schizandronic acid,

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Scheme 2. Hypothetical Biogenetic Pathway of 1



after many steps of oxidation, ring-opening, esterification, and possibly a free-radical mechanism to form the fivemembered carbon ring (ring B),⁸ as shown in Scheme 2.

Table 2.	Cytotoxic	Activity	of Co	mpounds	$1 - 4^{a}$
Labic 2.	Cytotoxic	retry	or co	mpounds	1 7

compounds	HL-60	SMMC-7721	A-549	SK-BR-3	PANC-1
1	>40	>40	17.7	5.2	5.9
1a	8.2	>40	36.6	34.3	>40
2	>40	>40	>40	23.5	4.1
3	>40	>40	>40	>40	>40
4	>40	>40	>40	>40	>40
cis-platin	1.6	13.6	11.8	19.9	14.4

^{*a*} Results are expressed as IC₅₀ values in micrometers, data were obtained from triplicate experiments, and *cis*-platin was used as positive control.

Compounds 1–4 were all tested for their cytotoxicity against HL-60, SMMC-7721, A-549, SK-BR-3, and PANC-1 human cancer cell lines. Among them, compounds 1 and 2

showed strong cytotoxicity against PANC-1 cell lines with IC₅₀ values of 5.9 and 4.1 μ M, respectively. Compound 1 also showed significant cytotoxicity against SK-BR-3 cell lines with an IC₅₀ value of 5.2 μ M (Table 2).

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Supporting Information Available: Detailed method of cytotoxicity test, experimental section, and 1D and 2D NMR spectra, UV, IR, optical rotation, and ¹H NMR data assignment of compounds 1-3. This material is available free of charge via the Internet at http://pubs.acs.org.

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